

1

2

3

4

5

6

7

0

1
<b>H</b>
hydrogen
1

## Key

relative atomic mass
<b>atomic symbol</b>
name
atomic (proton) number

4
<b>He</b>
helium
2

7	9
<b>Li</b>	<b>Be</b>
lithium	beryllium
3	4

11	12	14	16	19	20
<b>B</b>	<b>C</b>	<b>N</b>	<b>O</b>	<b>F</b>	<b>Ne</b>
boron	carbon	nitrogen	oxygen	fluorine	neon
5	6	7	8	9	10

23	24
<b>Na</b>	<b>Mg</b>
sodium	magnesium
11	12

27	28	31	32	35.5	40
<b>Al</b>	<b>Si</b>	<b>P</b>	<b>S</b>	<b>Cl</b>	<b>Ar</b>
aluminium	silicon	phosphorus	sulfur	chlorine	argon
13	14	15	16	17	18

39 <b>K</b> potassium 19	40 <b>Ca</b> calcium 20	45 <b>Sc</b> scandium 21	48 <b>Ti</b> titanium 22	51 <b>V</b> vanadium 23	52 <b>Cr</b> chromium 24	55 <b>Mn</b> manganese 25	56 <b>Fe</b> iron 26	59 <b>Co</b> cobalt 27	59 <b>Ni</b> nickel 28	63.5 <b>Cu</b> copper 29	65 <b>Zn</b> zinc 30	70 <b>Ga</b> gallium 31	73 <b>Ge</b> germanium 32	75 <b>As</b> arsenic 33	79 <b>Se</b> selenium 34	80 <b>Br</b> bromine 35	84 <b>Kr</b> krypton 36
85 <b>Rb</b> rubidium 37	88 <b>Sr</b> strontium 38	89 <b>Y</b> yttrium 39	91 <b>Zr</b> zirconium 40	93 <b>Nb</b> niobium 41	96 <b>Mo</b> molybdenum 42	[98] <b>Tc</b> technetium 43	101 <b>Ru</b> ruthenium 44	103 <b>Rh</b> rhodium 45	106 <b>Pd</b> palladium 46	108 <b>Ag</b> silver 47	112 <b>Cd</b> cadmium 48	115 <b>In</b> indium 49	119 <b>Sn</b> tin 50	122 <b>Sb</b> antimony 51	128 <b>Te</b> tellurium 52	127 <b>I</b> iodine 53	131 <b>Xe</b> xenon 54
133 <b>Cs</b> caesium 55	137 <b>Ba</b> barium 56	139 <b>La*</b> lanthanum 57	178 <b>Hf</b> hafnium 72	181 <b>Ta</b> tantalum 73	184 <b>W</b> tungsten 74	186 <b>Re</b> rhenium 75	190 <b>Os</b> osmium 76	192 <b>Ir</b> iridium 77	195 <b>Pt</b> platinum 78	197 <b>Au</b> gold 79	201 <b>Hg</b> mercury 80	204 <b>Tl</b> thallium 81	207 <b>Pb</b> lead 82	209 <b>Bi</b> bismuth 83	[209] <b>Po</b> polonium 84	[210] <b>At</b> astatine 85	[222] <b>Rn</b> radon 86
[223] <b>Fr</b> francium 87	[226] <b>Ra</b> radium 88	[227] <b>Ac*</b> actinium 89	[261] <b>Rf</b> rutherfordium 104	[262] <b>Db</b> dubnium 105	[266] <b>Sg</b> seaborgium 106	[264] <b>Bh</b> bohrium 107	[277] <b>Hs</b> hassium 108	[268] <b>Mt</b> meitnerium 109	[271] <b>Ds</b> darmstadtium 110	[272] <b>Rg</b> roentgenium 111	Elements with atomic numbers 112 – 116 have been reported but not fully authenticated						

\* The Lanthanides (atomic numbers 58 – 71) and the Actinides (atomic numbers 90 – 103) have been omitted.

Relative atomic masses for **Cu** and **Cl** have not been rounded to the nearest whole number.

# Forces

## Key Vocabulary:

**Acceleration:** The rate at which an object's velocity changes

**Air resistance:** The force of air acting on a moving object

**Balanced forces:** Two forces of equal size acting in opposite directions

**Contact force:** A force that must touch an object to affect it

**Friction:** The force caused by one surface touching another surface

**Gravity:** A force that attracts an object towards the centre of another object

**Magnetism:** The force between two magnets or between a magnet and a magnetic material

**Motion:** Movement

**Newton:** The unit for force

**Non-contact force:** A force that can affect an object without touching it

**Tension:** The force acting on an object that has been stretched

**Thrust:** A 'pushing' force

**Up-thrust:** The force that acts upwards on an object, often from air-resistance or water

**Velocity:** The scientific word for 'speed'

**Weight:** The force that results from an object's mass and the effect of gravity

## Energy Stores

**Kinetic energy** – All moving things have this. The amount depends on the mass of the object and its speed.

**Internal energy** – All objects have this. If it is caused by the movement of the particles in the object, it is **THERMAL ENERGY**. If it is due to how the particles are bonded together, it is **CHEMICAL ENERGY**.

**Elastic potential energy** – This is energy stored in stretched or squashed materials.

**Gravitational potential energy** – This is the energy an object has due to where it is positioned. It depends on the mass of the object, the height the object moves and the strength of gravity (the Gravitational field strength)

**Electrical energy** – Some objects carry electrical charges (called electrons). They can exert forces on each other.

**Magnetic energy** – Some objects can be magnetised and create magnetic fields. They can exert forces on other magnetised objects.

## Energy transfers

